



# NanoCat Group

[www.nanocat.co.in](http://www.nanocat.co.in)

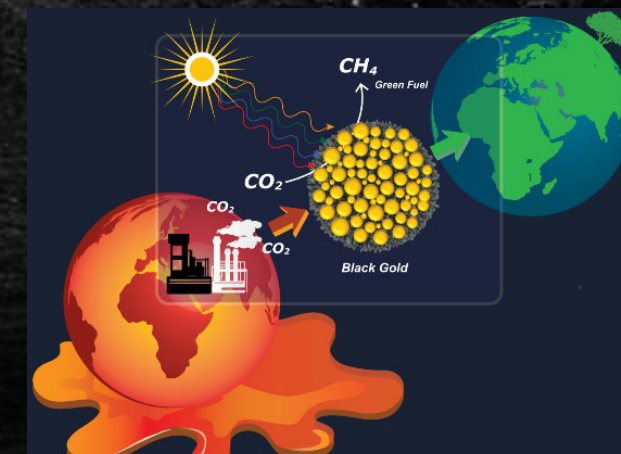


# Nanocatalysis to Combat Climate Change

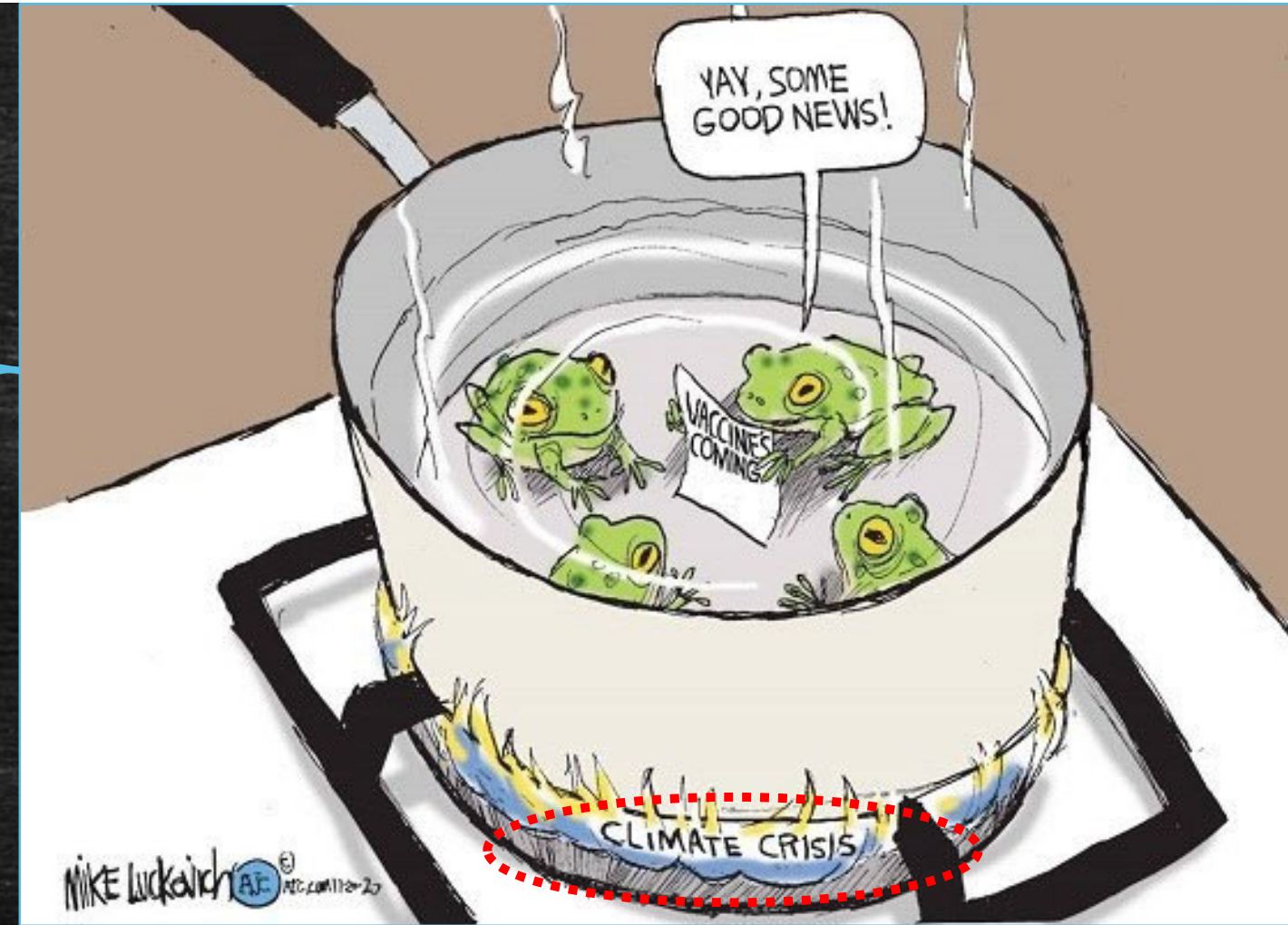


Vivek Polshettiwar

Tata Institute of Fundamental Research  
(TIFR), Mumbai



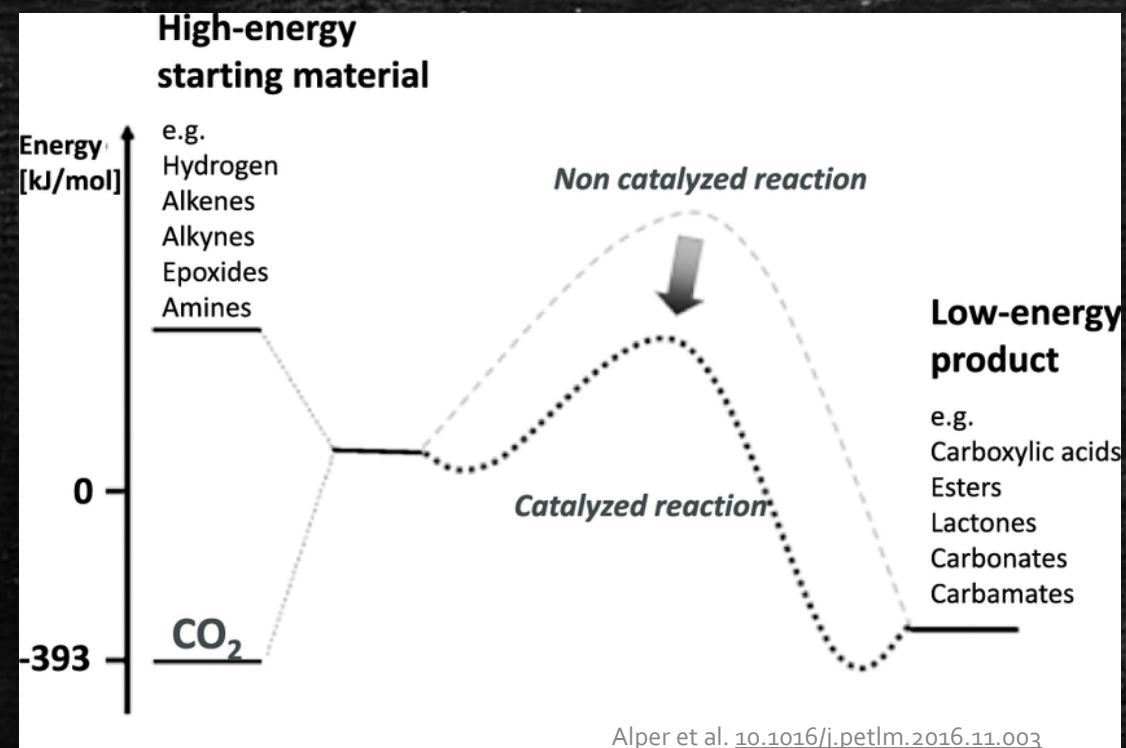
# **CLIMATE CHANGE** due to excessive carbon dioxide (CO<sub>2</sub>) is the “Most Serious Problem Mankind Has Ever Faced”.



*“The capture and utilization of carbon dioxide using solar energy and renewable hydrogen should become the preferred path to manufacture chemicals and fuels”*

# CO<sub>2</sub> Conversion is the Challenging Process

**CATALYSTS & NANOMATERIALS** can help reduce activation energy barrier & also harvest solar energy



**“STORING SUN ENERGY IN CO<sub>2</sub>”**

“CO<sub>2</sub> conversion to fuel using renewable H<sub>2</sub> can provide a solution to these two problems of,

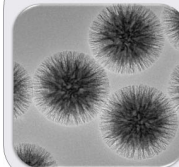
- i) excessive CO<sub>2</sub> levels, and
- ii) the temporal mismatch between renewable electricity production and demand as well as H<sub>2</sub> storage”



## Nanocatalysis Laboratory (to Combat Climate Change)



### Dendritic Fibrous Silica Nano-Silica (DFNS)

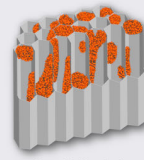


- Nano-Silica (120 to 1200 nm)
- High surface area (up to 1200 m<sup>2</sup>/g)
- High thermal stability (up to 950 °C)
- Hydro-thermal stability
- Excellent mechanical stability
- Fibrous surface morphology



DFNS

Vs

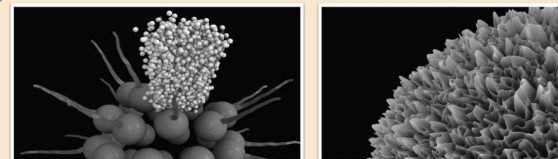


MCM-41/SBA-15

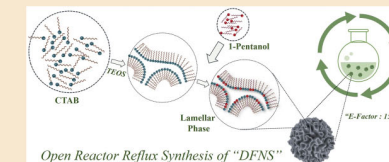
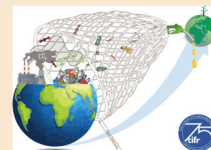
### Accessibility of Active Sites



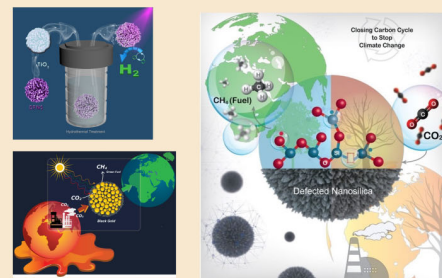
### Formation Mechanism, Size and Fiber Density Control of DFNS



- Dendritic Fibrous NanoSilica (DFNS)
- NanoTitania
- NanoTantalumOxide
- NanoCarbon
- NanoSilicaAluminate

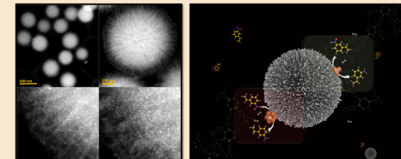


### Nanomaterials for CO<sub>2</sub> Capture & Conversion

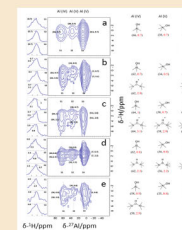
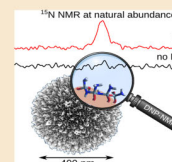


- Making DFNS Photoactive by Coating of TiO<sub>2</sub> on its Fibers using ALD and Solution Phase Protocols
- Use of DFNS as Hard Template for the Synthesis of High Surface Area Nanostructured Metal Oxides
- Design of Plasmonic Nano-Catalysts for Harvesting Solar Energy
- Hot Spot Engineering by Tuning Interparticle Plasmon Coupling
- CO<sub>2</sub> Methanation using Black Gold
- Defect Catalysed CO<sub>2</sub> Conversion

### Morphology Controlled Nanocatalysts



- Insights into the Catalytic Activity of Nitrated Fibrous Silica
- Tuning the Activity of SBA-15-Oxynitrides Solid Bases
- Sustainable Protocols for Supported Metal Nanocatalysts
- Ultrasmall NPs and Pseudo Single Atoms of Pt and Au based Catalysts with High TON and FOM
- Silica-BN for Catalytic Propane to Propene
- Aluminosilicates and Ironsilicates for C-H Activation



### Selected Publications from our Group

*ACS Sus. Chem. Eng.* **2013**, *1*, 1192  
*Angew. Chem. Int. Ed.* **2015**, *54*, 2190  
*Angew. Chem. Int. Ed.* **2015**, *54*, 5985  
*J. Mat. Chem. A.* **2016**, *4*, 7005; **2016**, *4*, 12416  
*ACS Catal.* **2016**, *6*, 2770

*Green Chem.* **2016**, *18*, 5890,  
*ChemPlusChem* **2016**, *81*, 1142  
*ChemSusChem*, **2017**, *10*, 2182 & *10*, 3866  
*J. Mat. Chem. A.* **2017**, *5*, 1935 & *5*, 14914  
*J. Phy. Chem A.* **2017**, *121*, 8080  
*Langmuir* **2017**, *33*, 13774

*Nature Prot.* **2019**, *11*, 5365  
*Chemical Sci.* **2019**, *10*, 6694  
*NanoScale*, **2019**, *11*, 5365  
*Proc. Natl. Acad. Sci.* **2020**, 6383  
*Nature Comm.* **2020**, *11*, 3828  
*Langmuir* **2020**, *36*, 12755  
*ACS Materials Lett.* **2020**, *2*, 699  
*ACS Sus. Chem. Eng.* **2020**, *8*, 16124  
*ACS App. Eng. Mater.* **2020**, *3*, 8150

*ACS App. Mat. Int.* **2018**, *10*, 23392  
*ACS App. Nano Mater.* **2018**, *1*, 3636  
*ChemPhotoChem*, **2018**, *2*, 796  
*ChemNanoMat* **2018**, 1231  
*ChemCatChem*, **2018**, 881  
*Chemical Science*, **2021**, *12*, 5774.  
*Chemical Science*, **2021**, *12*, 4825.  
*Chemical Comm.*, **2021**, *57*, 2005  
*Langmuir* **2021**, *37*, 6423  
*ACS Materials Lett.* **2021**, *3*, 574

### Group Members

Prof. Vivek Polshettiwar  
 Ms. Sushma Kundu  
 Mr. Rishi Verma  
 Mr. Rajesh Belgamwar  
 Mr. Saideep Singh  
 Ms. Gunjan Sharma  
 Mr. Pratip Chatterjee

### "Guiding Principles"

Catalytic performance (activity, kinetics, selectivity & stability) can be controlled by tuning the active sites (type, size, environment), defects and morphology of nanomaterial support.

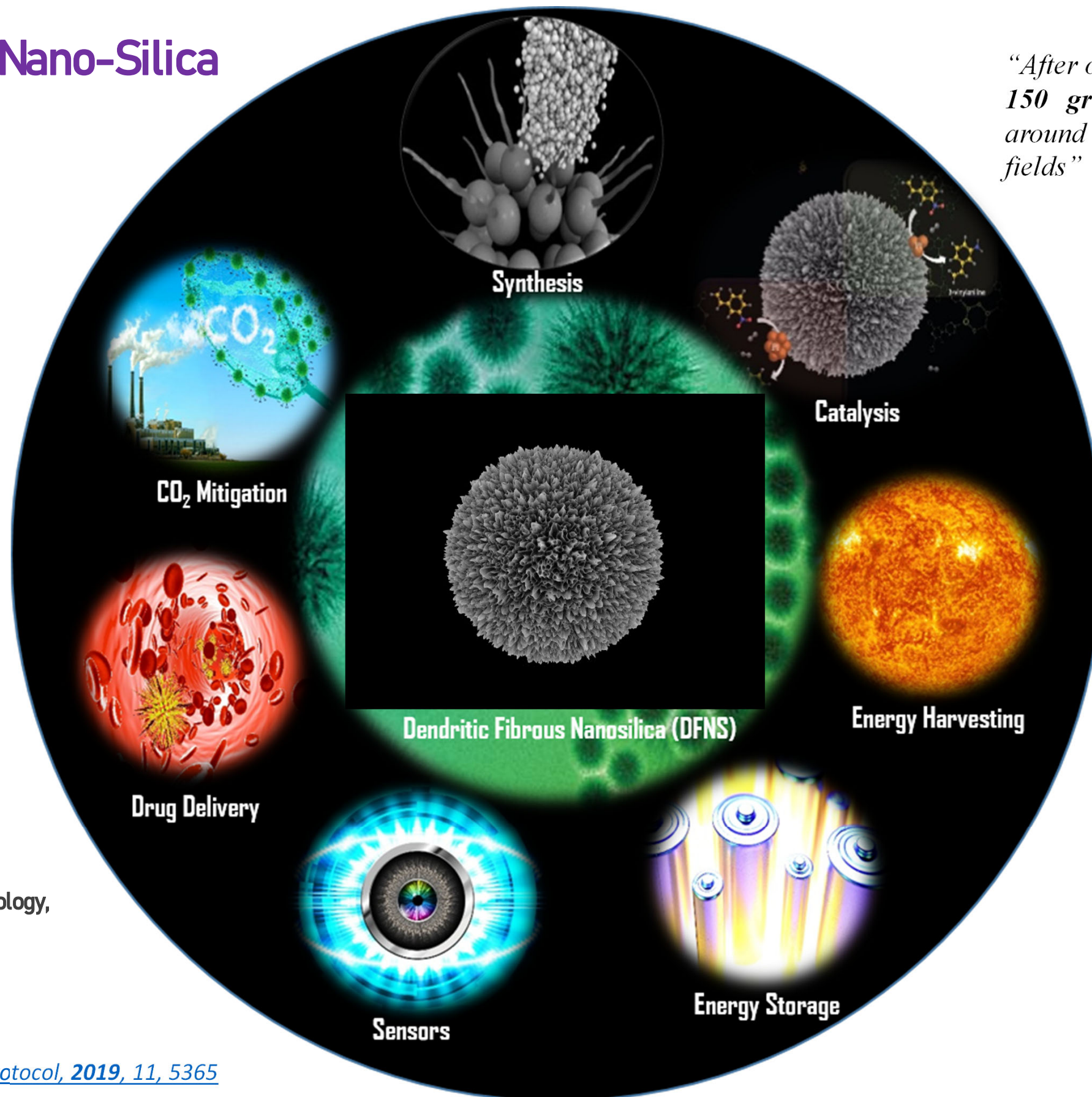


# Some of the Breakthroughs from Our NanoCat Lab

# Dendritic Fibrous Nano-Silica (DFNS)

“Indian Invented  
Material used by the  
World”

- Nano-Silica (50 to 1200 nm),
- High Surface Area (500 to 1200 m<sup>2</sup>/g),
- Thermal Stability (up to 800°C),
- Hydro-Thermal Stability,
- Mechanical Stability (up to 130 MPa),
- Unique Fibrous Surface Morphology,
- Easy to Synthesize.



“After our discovery of DFNS, now more than 150 groups worldwide using DFNS, with around 500 reports of use of DFNS in various fields”

**Polshettiwar et al.**

*Angew. Chem. Int. Ed.* **2010**, 49, 9652, *Angew. Chem. Int. Ed.* **2011**, 50, 2747,  
*Chemical Science*, **2012**, 3, 2224,  
*ACS Catalysis* **2012**, 2, 1425;  
*ACS Sus. Chem. Eng.* **2013**, 1, 1192,  
*Angew. Chem. Int. Ed.* **2015**, 54, 2190, *Angew. Chem. Int. Ed.* **2015**, 54, 5985,  
*J. Mat. Chem. A*, **2016**, 4, 7005; **2016**, 4, 12416;  
*ACS Catal.* **2016**, 6, 2770, *Green Chem.* **2016**, 18, 5890, *ChemPlusChem* **2016**, 81, 1142,  
*ChemSusChem*, **2017**, 10, 2182 & 10, 3866, *J. Mat. Chem. A*, **2017**, 5, 1935 & 5, 14914  
*J. Phy. Chem. A.* **2017**, 121, 8080; *Langmuir* **2017**, 33, 13774.  
*ACS App. Mater. Inter.* **2018**, 10, 23392; *ACS App. Nano Mater.* **2018**, 1, 3636;  
*ChemPhotoChem*, **2018**, 2, 796; *ChemNanoMat* **2018**, 1231; *ChemCatChem*, **2018**, 881.  
*Nature Protocol* **2019**, 11, 5365,  
*Chemical Science* **2019**, 10, 6694;  
*NanoScale* **2019**, 11, 5365;  
*Proc. Natl. Acad. Sci. U.S.A (PNAS)* **2020**, 6383;  
*Nature Comm.* **2020**, 11, 3828.  
*Chemical Science* **2021**, 12, 4285;  
*Chemical Science* **2021**, 12, 4267;  
*Chemical Science* **2021**, 5774.

*Indian Patents*

2016-21004089  
2020-21001441  
2020-21001440  
2020-21008717  
2020-21040554

*International Patents*

US Patent US20110253643  
PCT/IB2016/051870  
PCT/IN2020/050458

[Maity, Polshettiwar, et al. Nature Protocol, 2019, 11, 5365](#)

**SHINE ON BLACK GOLD FOR A BRIGHTER FUTURE**

When exposed to light, gold (Au) nanoparticles (NPs) generate catalysing "hot" electrons and "hotspots"

Au NP

"Hot" electron

"Hotspot"

Colloidosome

Black (nano)gold can be used for solar energy harvesting

To improve the catalyst's efficiency, "hotspots" in colloidosomes can be tuned by tuning the size and gaps between Au NPs

- CO<sub>2</sub> conversion
- Fuel CH<sub>4</sub>
- Protein unfolding
- Aldehyde hydrosilylation
- Alcohol oxidation
- Seawater desalination

The use of black gold can help us move one step closer to combating climate change

Chemical Science

PICK OF THE WEEK

Plasmonic colloidosomes of black gold for solar energy harvesting and hotspots directed catalysis for CO<sub>2</sub> to fuel conversion  
Dhiman, Maity et al. (2019) | DOI: 10.1039/C9SC02369K

ROYAL SOCIETY OF CHEMISTRY

# Black Gold : an Artificial Tree

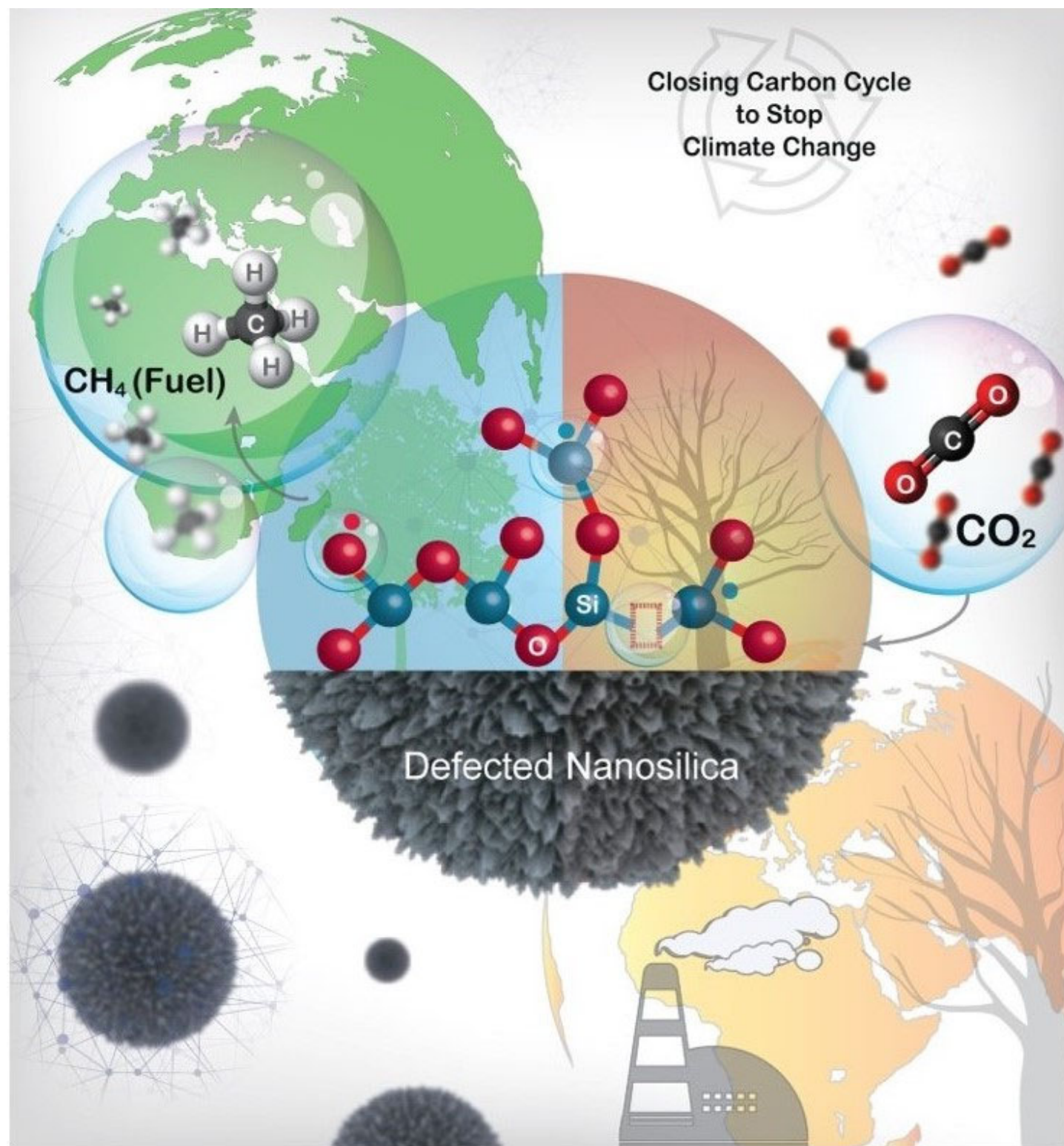
Black Gold absorbs the entire visible and near-infrared region of solar light, due to interparticle plasmonic coupling.

Black Gold acts like an artificial tree that uses CO<sub>2</sub>, sunlight and water to produce fuel.  
It also converts sea water to potable water.

*Polshettiwar et al. Chemical Sci. 2019, 10, 6694*



Mr. Mahak Dhiman



# “Perfecting Imperfection”

for  $\text{CO}_2$  Conversion

**Defects in Nanosilica Catalytically Convert  $\text{CO}_2$  to Methane**

First example of Metal-Free-Ligand-Free  $\text{CO}_2$  Conversion

Process



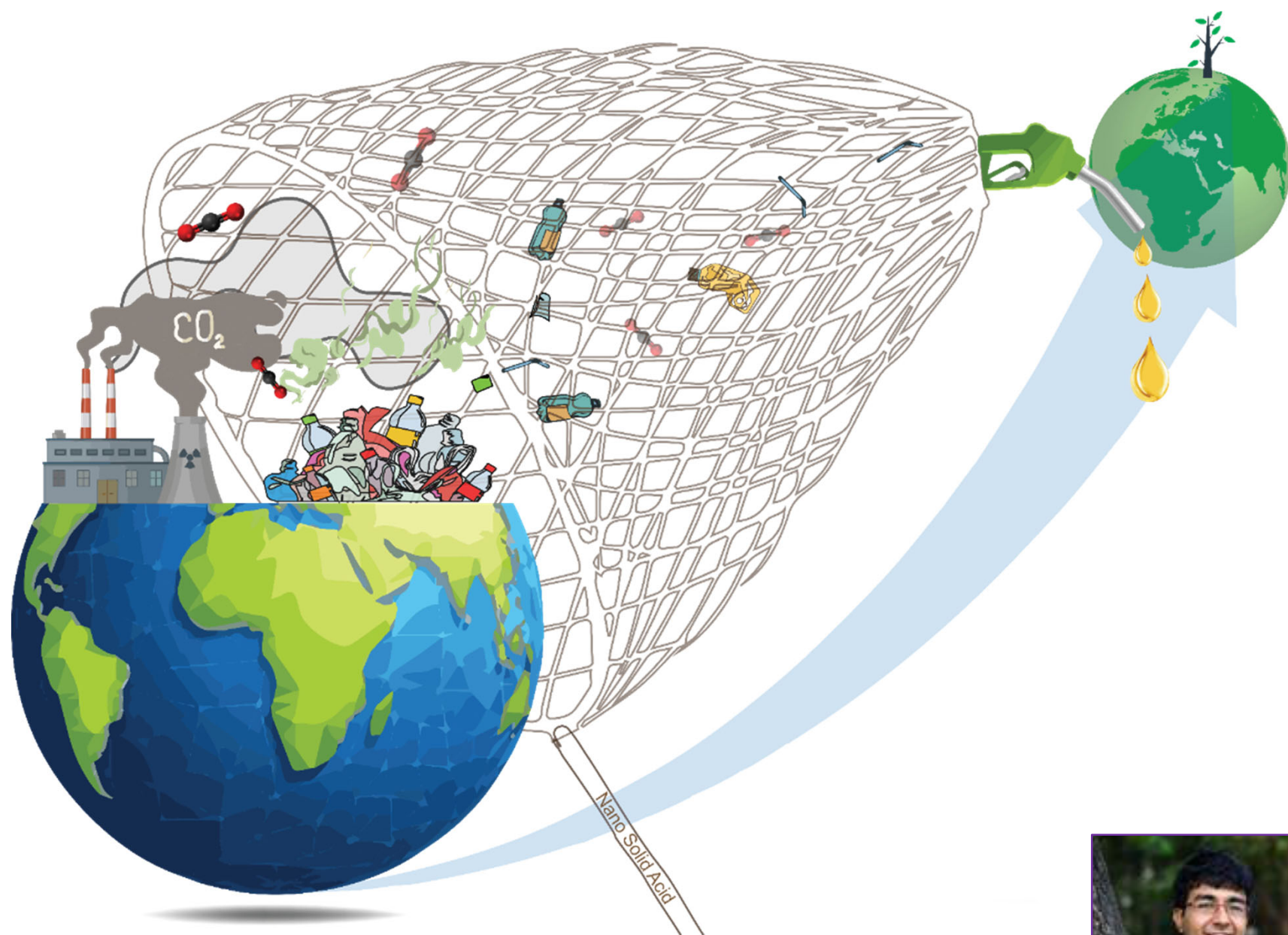
Mr. Amit Mishra

*Polshettiwar et al. Proc. Natl. Acad. Sci. U.S.A (PNAS) 2020, 117, 6383.*



# “Solid Acids for Waste Plastic and CO<sub>2</sub> Chemicals”

Catalytic Nanosponges of Acidic Aluminosilicates for Plastic Degradation and CO<sub>2</sub> to Fuel Conversion.



Alkanes C5-C21 chain length (53%)

C2-C4 alkenes (46.6%)

Coke (0.4%)



Mr. Ayan Maity

## “DNFS based Sorbents for CO<sub>2</sub> Capture”

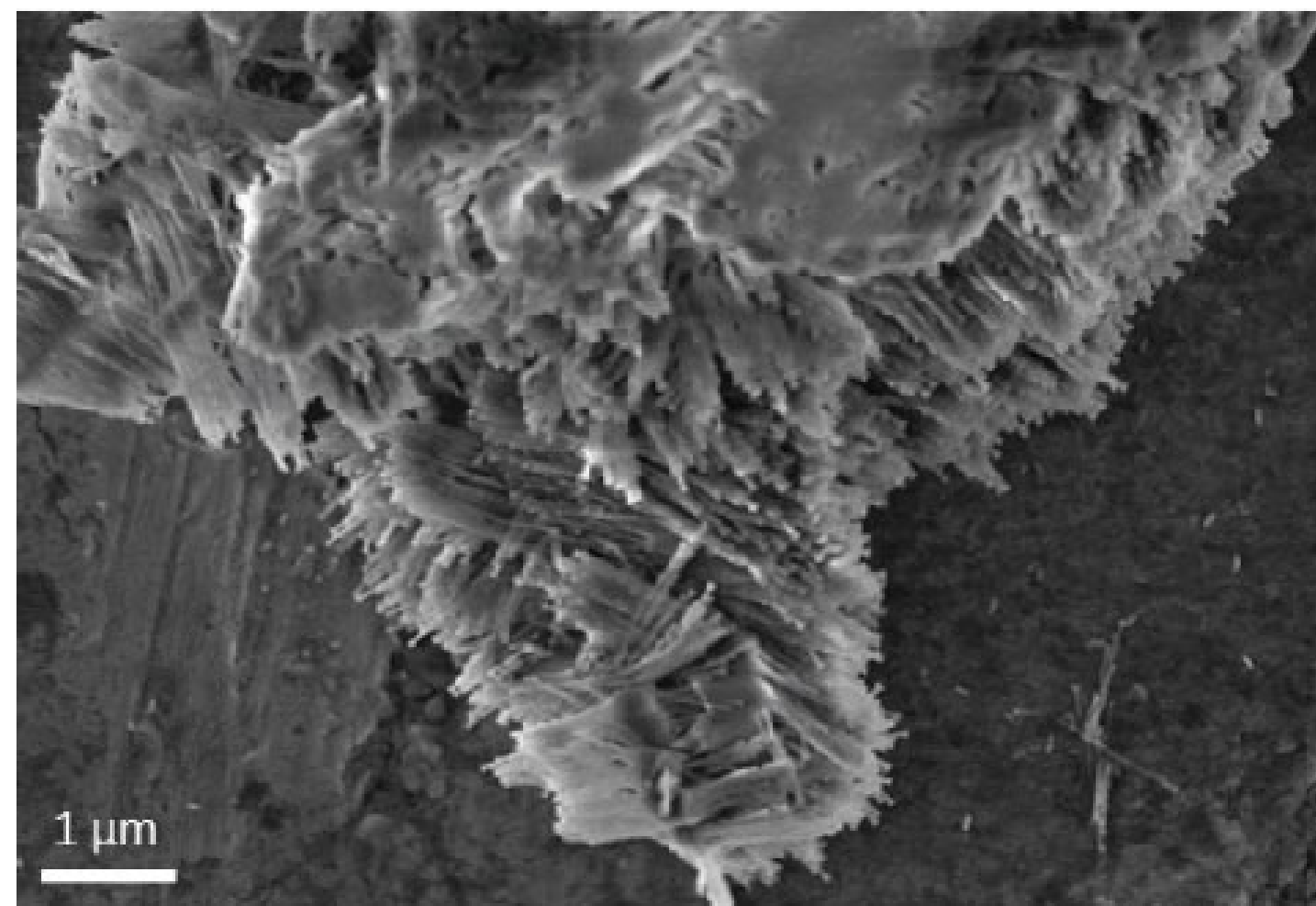
Capture Good Amount of CO<sub>2</sub> at Moderate Temperature and Atmospheric Pressure. Faster Kinetics, Stable and Recyclable and Selective.



[Polshettiwar,\\* et al. \*Chem. Sci.\*, 2012, 3, 4222; \*J. Mat. Chem. A\*, 2016, 4, 7005.](#)

## “High Temperature CO<sub>2</sub> Capture”

Lithium silicate nanosheets (LSNs) showed high CO<sub>2</sub> capture capacity with ultra-fast kinetics and enhanced stability, at least 200 cycles.



[Belgamwar, Polshettiwar\\* et al. \*Chemical Science\*, 2021, 12, 4825.](#)

## CO<sub>2</sub> Conversion – “Dream Process”

- at ambient temperature and pressure
- without external energy
- water as a hydrogen source
- using an inexpensive and abundant material

# Transforming $\text{CO}_2$ and Sea Water to Green **HYDROGEN** & Green **CEMENT** Using Magnesium Scrap

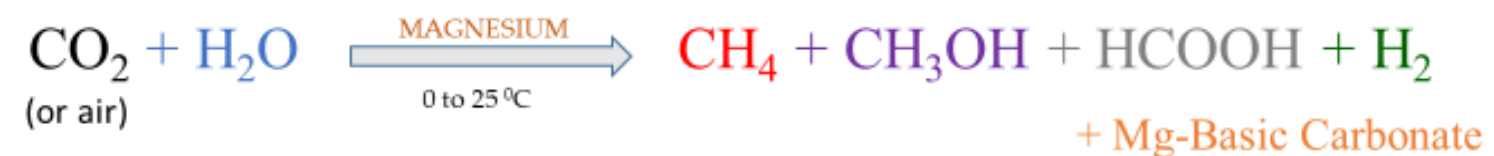


## “CO<sub>2</sub> to Fuel and Cement at Room Temperature”

Bubble the Air in Water with a pinch of Magnesium and you will get Fuel and Green Cement, at room temperature and pressure.



## “Process for CO<sub>2</sub> Conversion”

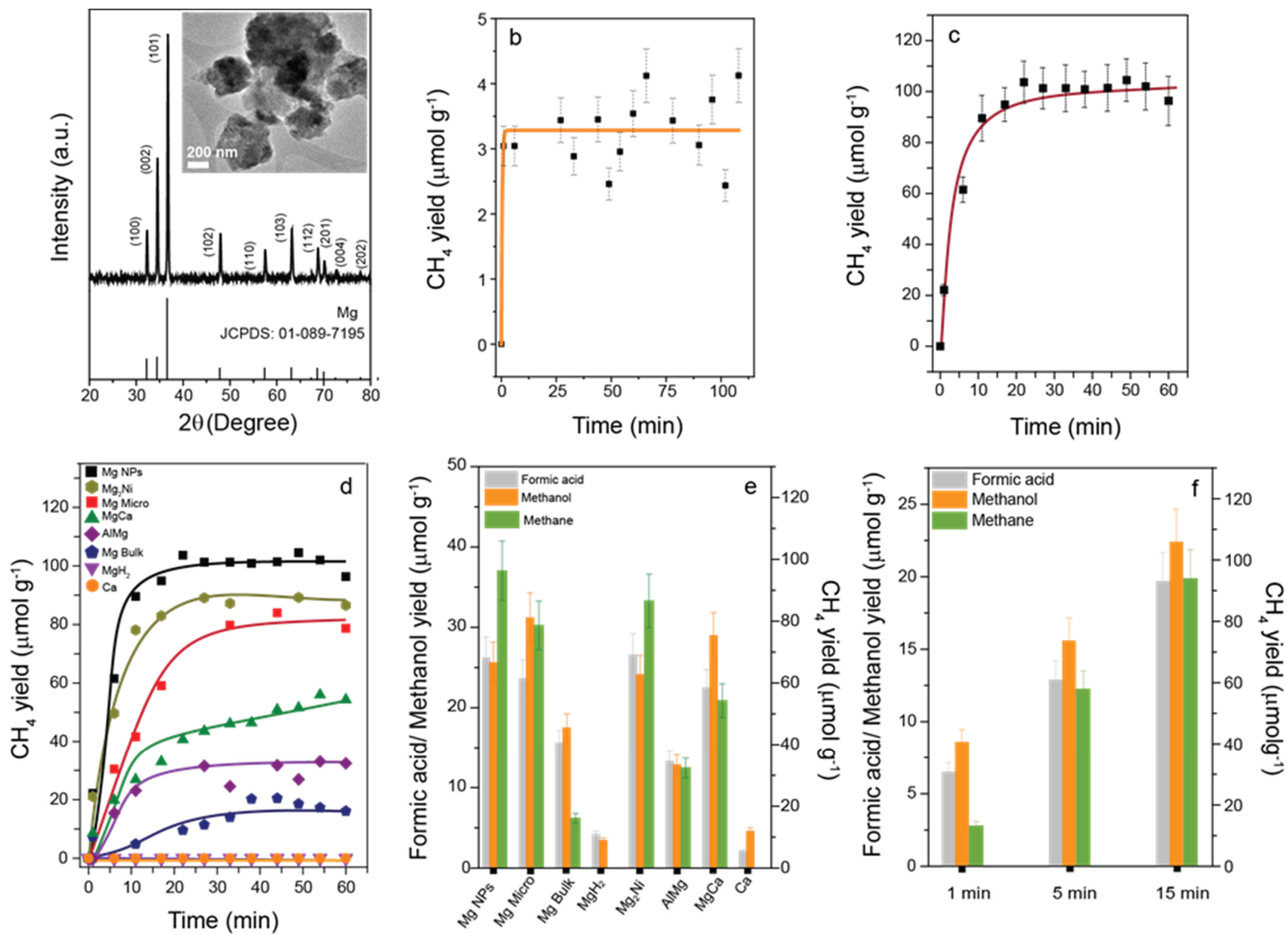


*A direct conversion of CO<sub>2</sub> to fuel,*

- at room temperature and atmospheric pressure,
- without the need for any external energy (*no light, no electricity, no heat*),
- just using water (*no external hydrogen, even sea water works*).

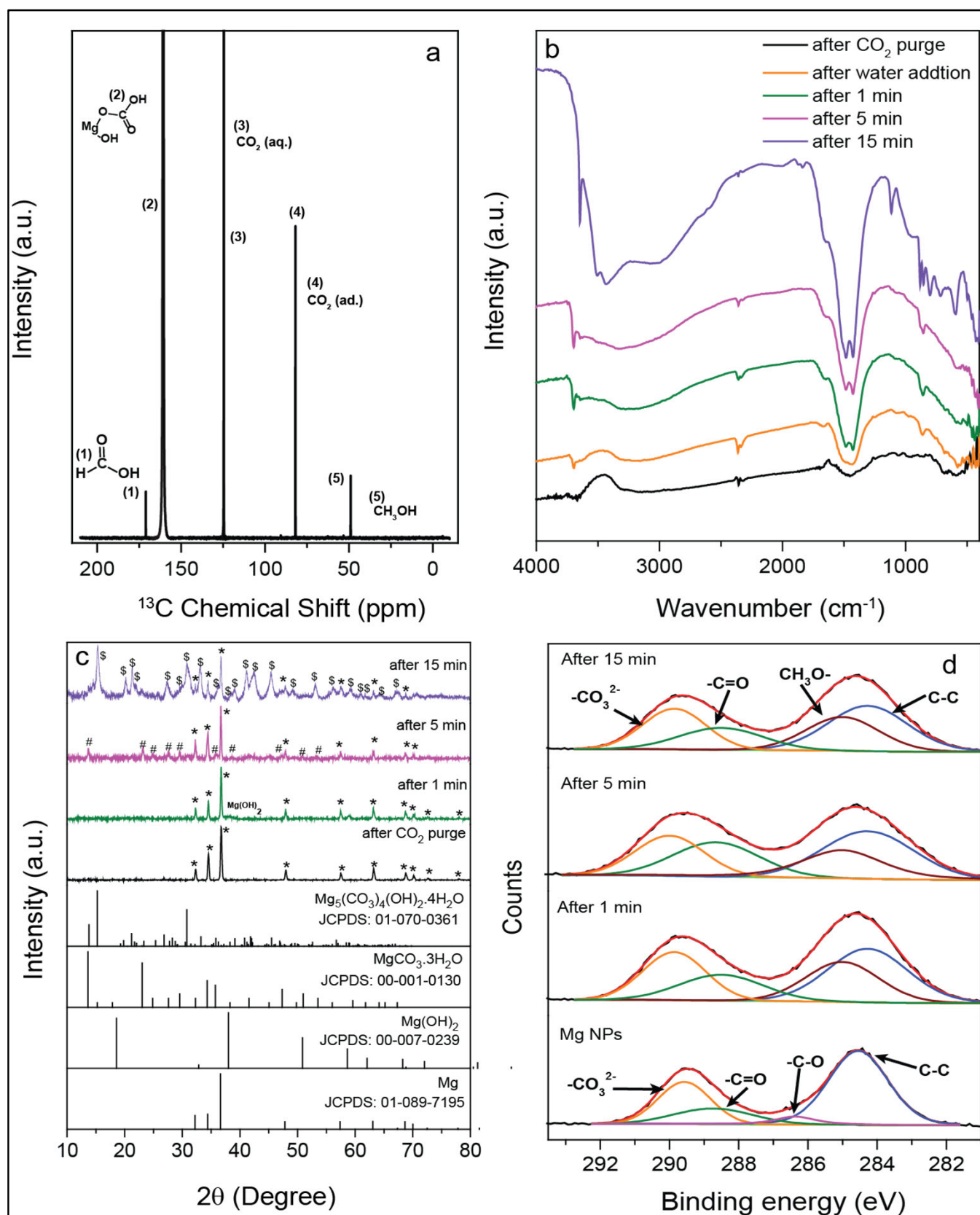
Magnesium is the 4<sup>th</sup> most common element in the Earth (after iron, oxygen and silicon), making up 13% of the earth's mass.

“CO<sub>2</sub> will be durably sequestered for more than 100 year in the form of **Green Cement**”

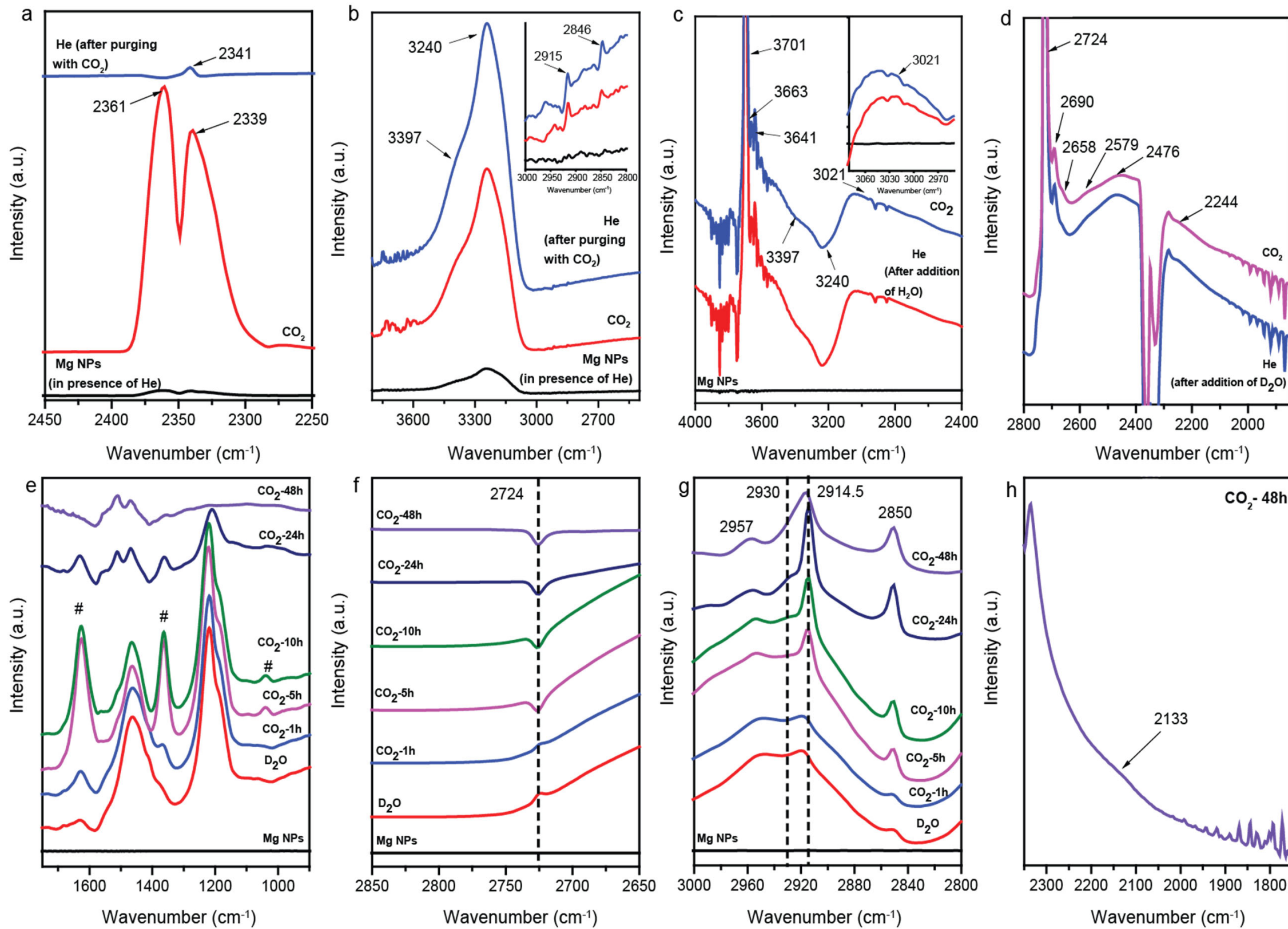


**Figure 1.** (a) PXRD pattern (TEM image as inset) of Mg NPs along with standard XRD pattern of Mg;

Methane yield with reaction time for CO<sub>2</sub> reduction by Mg NPs using (b) CO<sub>2</sub> directly from the air and (c) 99.99% CO<sub>2</sub>



- (a)  $^{13}\text{C}$  NMR spectrum of the reaction mixture (using labeled  $^{13}\text{CO}_2$ ) after 1 h of reaction on Mg NPs;
- (b) FTIR spectra,
- (c) XRD patterns and
- (d) XPS C 1s spectra of solid products recovered after 1, 5, and 15 min of the reaction of  $\text{CO}_2$  with Mg NPs and water, at room temperature and atmospheric pressure.



In situ **ATR-FTIR** spectra:

(a) and (b) in He, CO<sub>2</sub>, and again He,

(c) after adding H<sub>2</sub>O to the Mg NPs and subsequently purging with He and CO<sub>2</sub> gas,

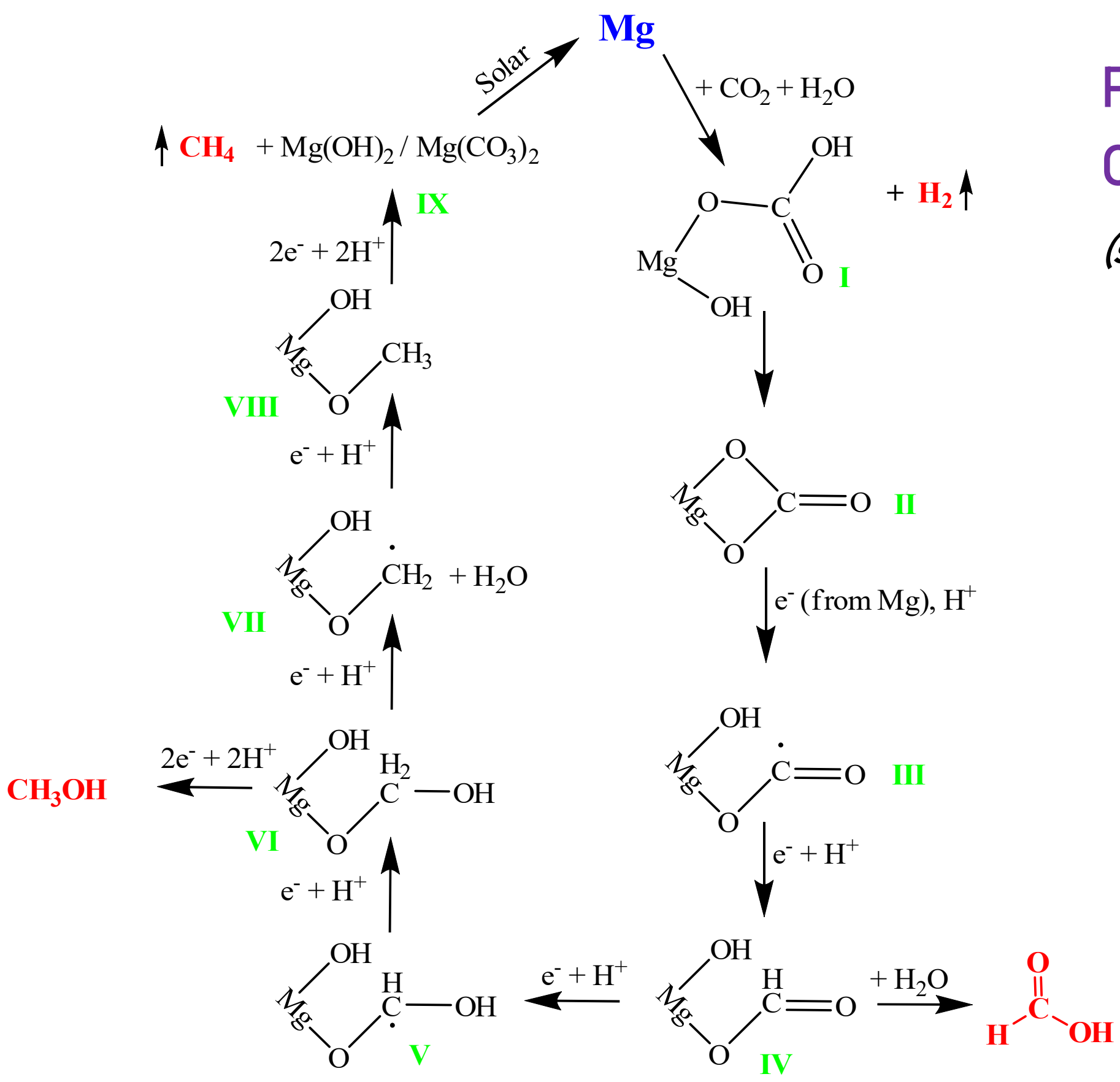
d) after adding D<sub>2</sub>O to the Mg NPs and subsequently purging with He and CO<sub>2</sub> gas,

(e-h) after adding D<sub>2</sub>O to the Mg NPs and then purging with CO<sub>2</sub> for 1, 5, 10, 24 and 48 h.

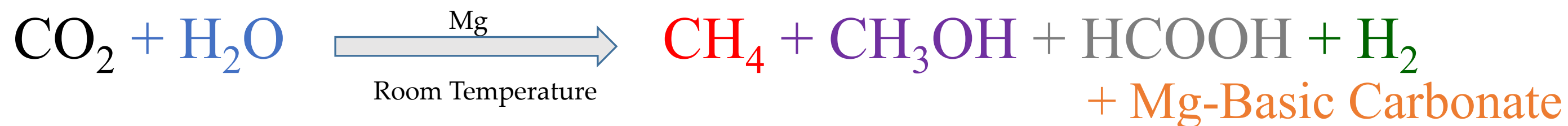


# Plausible reaction mechanism of CO<sub>2</sub> conversion by Mg using water

(supported by *in-situ* ATR and DFT)



# “Process Sustainability”



1 kg of Mg via simple reaction with water and CO<sub>2</sub> produces in 1 hr,

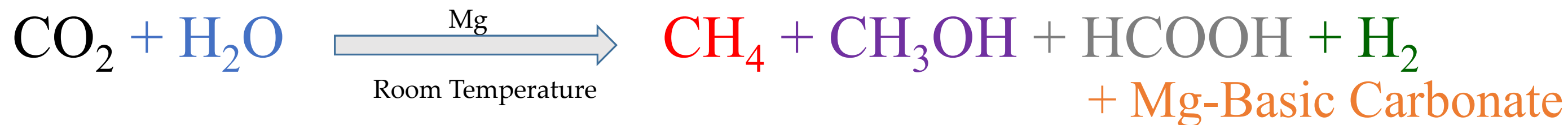
- Methane (2.43 lit),
- Hydrogen (940 lit),
- Mg-basic carbonate (3.85 kg),
- Small amount of methanol, and formic acid

## # Mg Basic Carbonate

Mg-Basic Carbonate Uses:

- **Green Cement**
- Ancient building material
- Pharmaceuticals (e.g. antacid and laxatives)
- Flame retardant
- Colour & paints Industries (Fillers for paints, textiles, plastics, etc)

# “Cost per Tonne”



**1 kg of Mg via simple reaction with water and CO<sub>2</sub> produces in 1 hr,**

- **Methane** (2.43 lit),
- **Hydrogen** (940 lit),
- **Mg-basic carbonate** (3.85 kg),
- Small amount of methanol, formic acid

Price Calculations	Input Cost	Product Cost
1 kg <b>Mg</b> (SCRAP)	<b>USD 2</b>	-
3.85 kg <b>Mg-Basic Carbonate</b> (Current Cost is USD 26)	-	<b>USD 100</b> (USD 26 x 3.85)
2.43 lit Methane	-	<b>Green</b>
940 lit Hydrogen	-	<b>Green</b>
Total Profit Per Kg Per ton		USD 98 USD 98000
2.2 kg CO <sub>2</sub> mitigated per kg of Magnesium		

“Life-Cycle Assessment (LCA)” Preliminary (not comprehensive)

Process	Energy (+ required or - produced)	CO <sub>2</sub> (+ released or - mitigated)	
Electrolysis of Mg <sup>2+</sup> to produce 1 Kg Mg	+ 0 MJ	+ 0 kg	
	Use of Magnesium SCRAP	Use of Magnesium SCRAP	
CO <sub>2</sub> conversion by 1 kg Mg	No external energy was required for this reaction	0 MJ	
	Product- <b>Methane</b> : Energy (50.1 kJ/g) released by burning 1.6 g of Methane	-0.0816 MJ 0.0044 kg (mitigated) 0.1 mol CO <sub>2</sub> converted to methane - 0.1 x 44 = 4.4 g	
	Product- <b>Methanol</b> : Energy (22.7 kJ/g) released by burning 0.8138 g of Methanol	-0.0185 MJ 0.0011176 kg (mitigated) 0.0254 mol CO <sub>2</sub> converted to methanol -0.0254 x 44 = 1.1176 g	
	Product- <b>Formic Acid</b> : Energy (6.4 MJ/lit) released by burning 0.98 mL of Formic Acid	-0.00627 MJ 0.001196kg (mitigated) 0.026 mol CO <sub>2</sub> converted to formic acid-0.026 x 46 = 1.196 g	
	Product- <b>Hydrogen</b> : Energy (0.490 MJ/kg) required to produce 0.083kg hydrogen from fossil fuel (steam reforming of methane), which is saved by this process	-0.0406 MJ - 0.76 kg (stopped from releasing to the environment) 9.21 kg CO <sub>2</sub> per kg H <sub>2</sub> emitted if produced from methane reforming, i.e. 0.76 kg CO <sub>2</sub> could have been emitted if H <sub>2</sub> was produced from methane.	
	Energy (141.86 MJ/kg) released by burning 0.083 kg hydrogen	-11.77 MJ 0 kg (no emission)	
	Product- <b>Mg Basic Carbonate</b> : Mg basic carbonate (heat of formation of carbonate)	-2.08 MJ - 1.44 kg (mitigated) 1 kg Mg produced 3.85 kg basic carbonate by reacting with 1.44 kg CO <sub>2</sub>	
	<b>Total</b>	<b>11.91 MJ (Energy generated)</b>	<b>2.2 kg (CO<sub>2</sub> mitigated)</b>

## Magnesium (Mg) SCRAP, Energy and Scalability

### Amount of Magnesium SCRAP needed

For capturing and converting 1000 tonnes of CO<sub>2</sub> per year (into 3850 tonnes of **GREEN CEMENT**),  
We need 690.79 tonnes of **Mg SCRAP** per year

### Availability of Magnesium SCRAP

In 2016 total magnesium consumption worldwide was 843kt,  
with production at **878kt**. (just **0.08 % scrap per year**)



<http://www.thementalcasting.com/magnesium-scrap.html>

### Scalability

Current scale: 1 gm

Simple mixing of Mg powder, water and CO<sub>2</sub> (air)  
at room temperature and atmospheric pressure

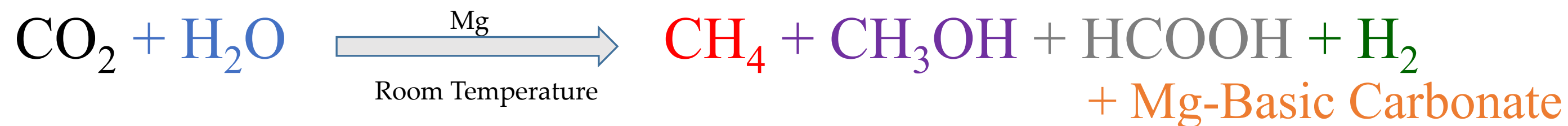
### Energy and Water

- without the need for any external energy  
(*no light, no electricity, no heat*),
- just using water  
(*even sea water works*).

### Waste to Wealth

“CO<sub>2</sub> will be durably sequestered for more than 100 year in  
the form of **Green Cement** using **scrap (waste) Magnesium**”

## “Process Sustainability”

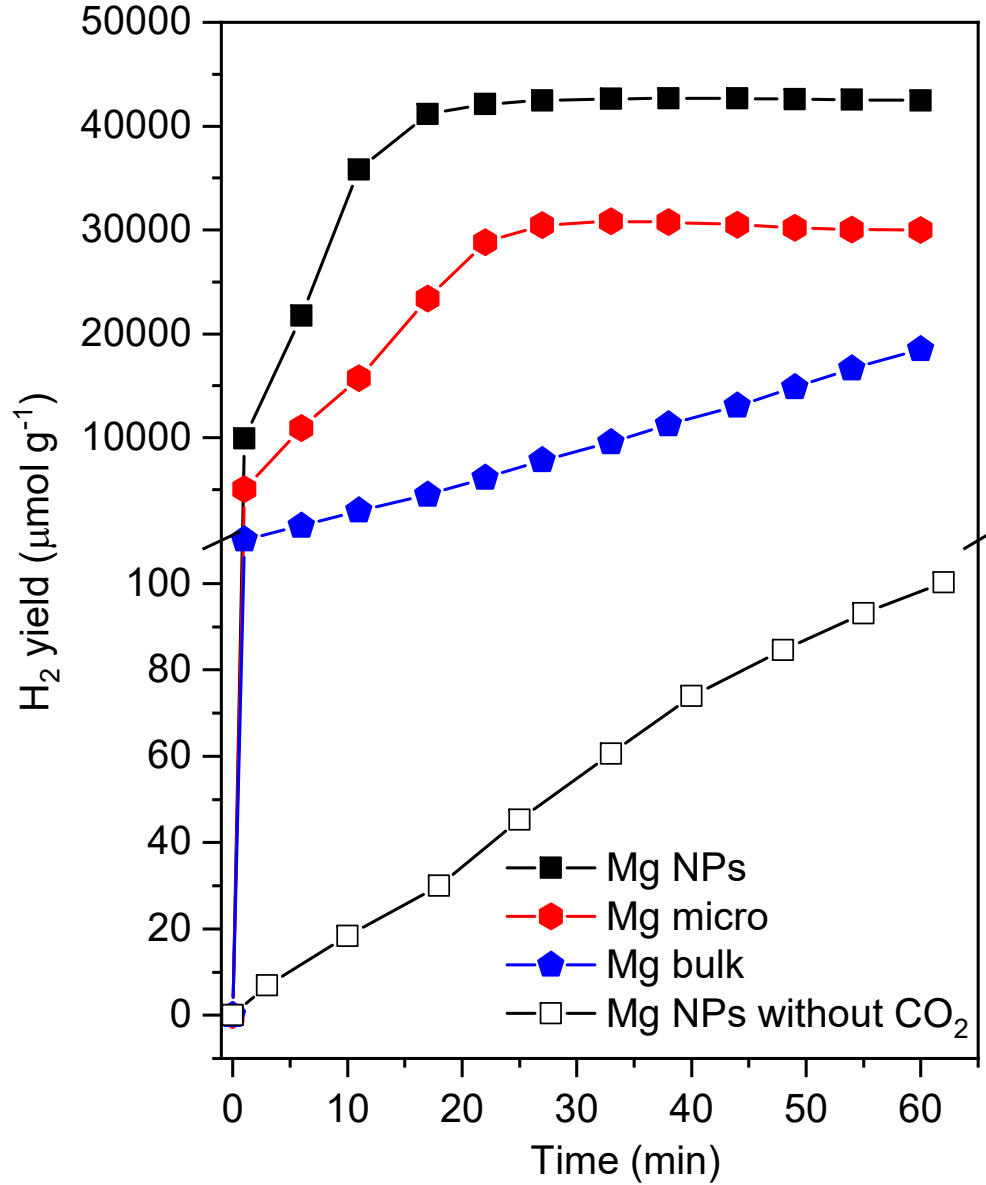


**1 kg of Mg via simple reaction with water and CO<sub>2</sub> produces in 1 hr,**

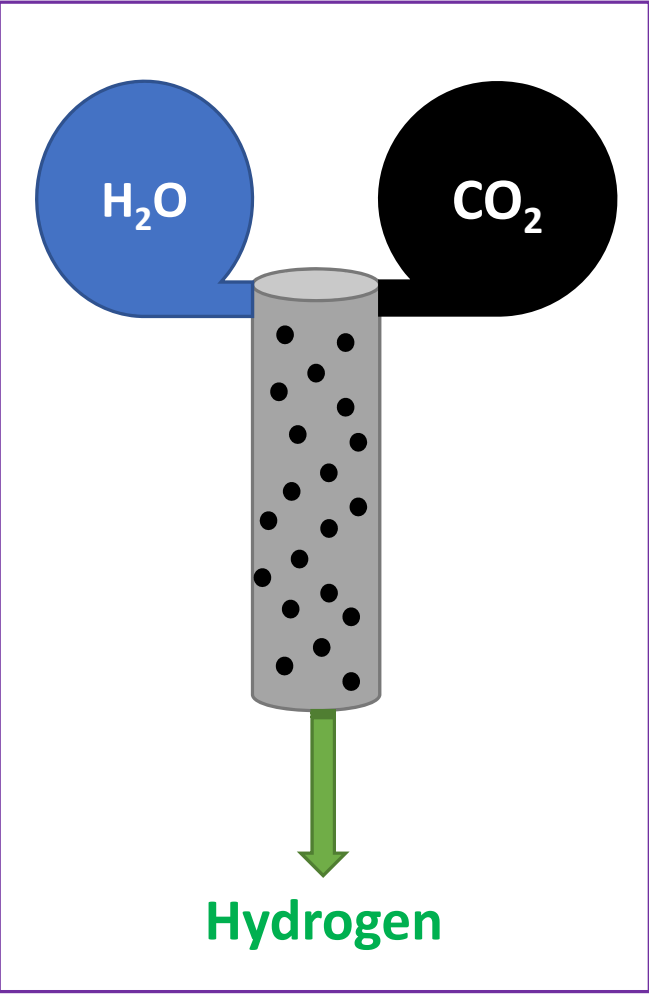
- **Methane** (2.43 lit),
- **Hydrogen** (940 lit),
- **Mg-basic carbonate** (3.85 kg),
- Small amount of methanol, and formic acid

**# Green Hydrogen**

Our Mg protocol can even be used for H<sub>2</sub> production (940 liter per kg of Mg), which is nearly 420 times more than H<sub>2</sub> produced by the reaction of Mg with water alone (2.24 liter per kg of Mg).



### “Green Hydrogen on Demand”



Portable Magnesium Cartridge which will produce H<sub>2</sub> from Water and CO<sub>2</sub> at Room Temperature and Atmospheric Pressure

# Green Hydrogen

# Price of Mg Metal

**indiamart** All India Enter product / service to search Search Get Best Price Covid-19 Supplies Sell Help Messages Sign In

IndiaMART > Metal & Metal Made Products > Magnesium Ingots

**Pure Magnesium Ingot**  
₹ 280/ Kilogram [Get Latest Price](#)

Material: Magnesium  
Country of Origin: Made in India  
Physical State: Solid  
Purity: 99.99%  
Boiling Point: 1090 Deg C  
Melting Point: 650 Deg C

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**Rs. 285 per Kg**

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**RAWS**

Low Price high purity Mg magnesium ingots magnesium metal  
FOB Reference Price: [Get Latest Price](#)

**\$500.00 - \$600.00** / Metric Ton | 5 Metric Ton/Metric Tons(Min. Order)

Dimensions: MG99.99% FE0.002% SI0.002% MN0.002%

Lead Time:

Quantity(Metric Tons)	1 - 500	>500
Est. Time(days)	15	To be negotiated

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Manufacturer,Trading Company  
CN 2 YRS  
≤11h Response Time  
100.0% On-time delivery rate

**Rs. 45 per Kg**

**Magnesium Ingot**  
₹ 980/ Ton [Get Latest Price](#)

Material: Magnesium  
Minimum Order Quantity: 50 Ton

Mn Ingot  
Keeping in mind the precise demands of clients, we supply a qualitative range of Magnesium Ingot to our valued clients, these are processed using best available technologies. Designed as per

View Complete Details

Fill the quantity to get latest price!  
Quantity  kg [Get Latest Price](#)  
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Verified Supplier Exporter  
Company Video  
Ask for more details from the seller  
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**Rs. 1 per Kg**

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Promotional Top Quality Factory Manufacture Various Metal Magnesium ingot  
FOB Reference Price: [Get Latest Price](#)

**\$180.00 - \$200.00** / Metric Ton | 50 Metric Ton/Metric Tons(Min. Order)

Weight: 7.5KG

Lead Time:

Quantity(Metric Tons)	1 - 1000	1001 - 20000	>20000
Est. Time(days)	10	30	To be negotiated

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Trading Company  
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≤24h Response Time  
100.0% On-time delivery rate

**Rs. 15 per Kg**

Price ~ Rs. 700 per Kg (after powder)



# Price of Mg Basic Carbonate (CAS Number: 39409-82-0)

https://www.sigmaaldrich.com

Product #	Description	SKU-Pack Size	Availability	Pack Size	Price (INR)	Quantity
13118	Magnesium carbonate basic, purum, light, ≥40% Mg (as MgO) basis, powder (light),	13118-1KG	✓ Available to ship on 10.03.2021 - FROM	1 kg	2,100.05	0
		<a href="#">Bulk orders?</a>		<a href="#">ADD TO CART</a>		
63062	Magnesium carbonate basic, purum p.a., heavy, ≥40% (MgO),	63062-1KG	✓ Available to ship on 10.03.2021 - FROM	1 kg	3,994.43	0
		<a href="#">Bulk orders?</a>		<a href="#">ADD TO CART</a>		
63032	Magnesium carbonate basic, tested according to Ph. Eur., heavy,	63032-1KG-F	✓ Estimated to ship on 22.03.2021 - FROM	1 kg	4,535.68	0
		<a href="#">Bulk orders?</a>		<a href="#">ADD TO CART</a>		

Price for Magnesium Carbonate ~ Rs. 100 per Kg (not basic)

Price ~ Rs. 2000 per Kg

# Looking for Experts as our Team Members

## 1. Reactor Design

(Batch to flow, pilot plant design)

## 2. Chemical Engineer

(Up-scalable Process Development, Process Flow Diagram, Stream Table, and Mass & Energy balance etc.)

## 3. LCA Analyst

(full cradle-to-grave LCA, Fully considered cost per tonne)

Please Contact: Prof. Vivek Polshettiwar

Emails:

[polshettiwar@gmail.com](mailto:polshettiwar@gmail.com)

[vivekpol@tifr.res.in](mailto:vivekpol@tifr.res.in)

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**“The outcome of basic research must help in the development of commercial products and technology to help resolve India’s societal challenges”**



India can not continue *importing oil, solar panels, batteries*

and

then in coming times

*CO<sub>2</sub> Capture and Conversion Technologies*, which is the future.

India imported 270 Million Metric Ton of **crude oil** valuing \$120 billion in 2019-20.

India bought **batteries** worth \$ 1.2 billion in 2019-20.

India imported **solar cells and modules** worth \$1,180 mn from China in Apr-Dec FY20



# VISION of Dr. Homi J. Bhabha

Dr. Homi J. Bhabha, one of the chief architects of Government of India's Scientific Policy Resolution (1958), said

“The key to national prosperity, apart from the spirit of the people, lies in the modern age, in the effective combination of three factors: technology, raw material and capital, of which the first is perhaps the most important, since the creation and adoption of new scientific techniques can, in fact, make up for a deficiency in natural resources, and reduce the demands on capital. But technology can only grow out of a study of science and its applications.”

Dr. Homi Bhabha built Department of Atomic Energy (DAE) and BARC and gave India **Nuclear Power**, the best example of Indigenous development and "Atmanirbhar Bharat".

Now it's time to harvest **Solar Power** and **CO<sub>2</sub>** for India's growth.



**Dr. Homi Jehangir Bhabha**

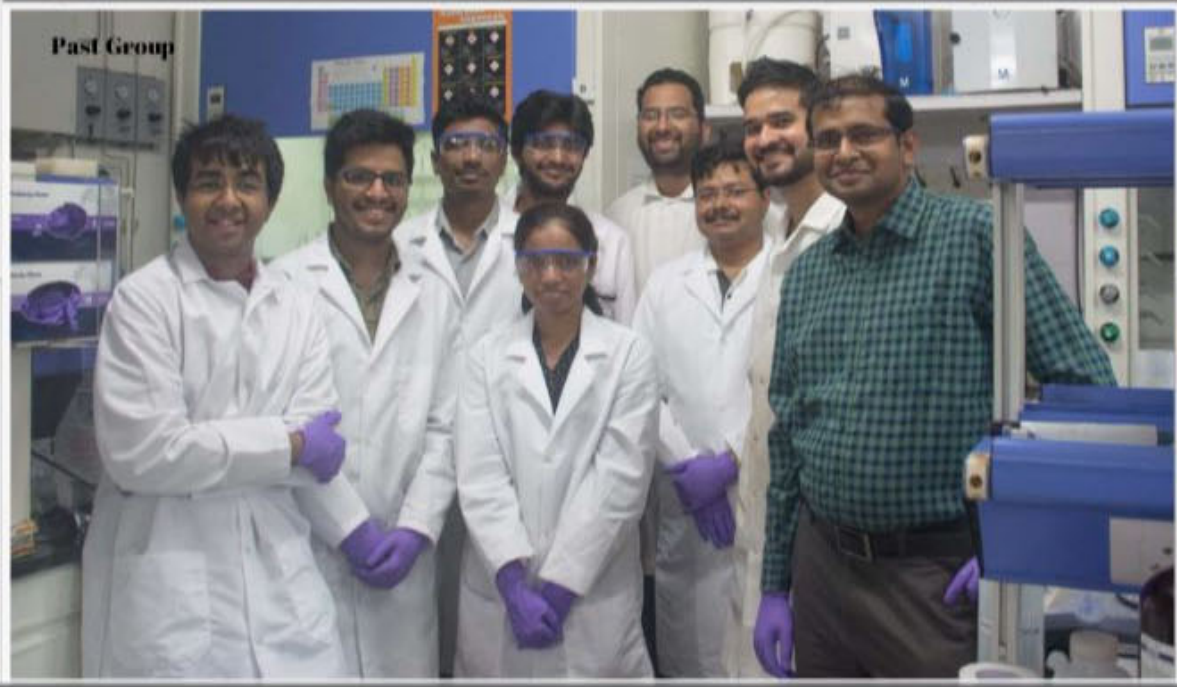
“The key to national prosperity, apart from the spirit of the people, lies, in the modern age, in the effective combination of three factors, technology, raw materials and capital, of which the first is perhaps the most important, since the creation and adoption of new scientific techniques can, in fact, make up for a deficiency in natural resources, and reduce the demands on capital. But technology can only grow out of the study of science and its applications”

~ Opening statement of the Government of India's Scientific Policy Resolution, 1958, of which Bhabha was one of the principal architects.




**Nanocatalysis Laboratories**  
for Energy & Environment

PI: Prof. Vivek Polshettiwar, Group Web: [www.nanocat.co.in](http://www.nanocat.co.in)



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हिन्दुस्तान पेट्रोलियम

Thank you !